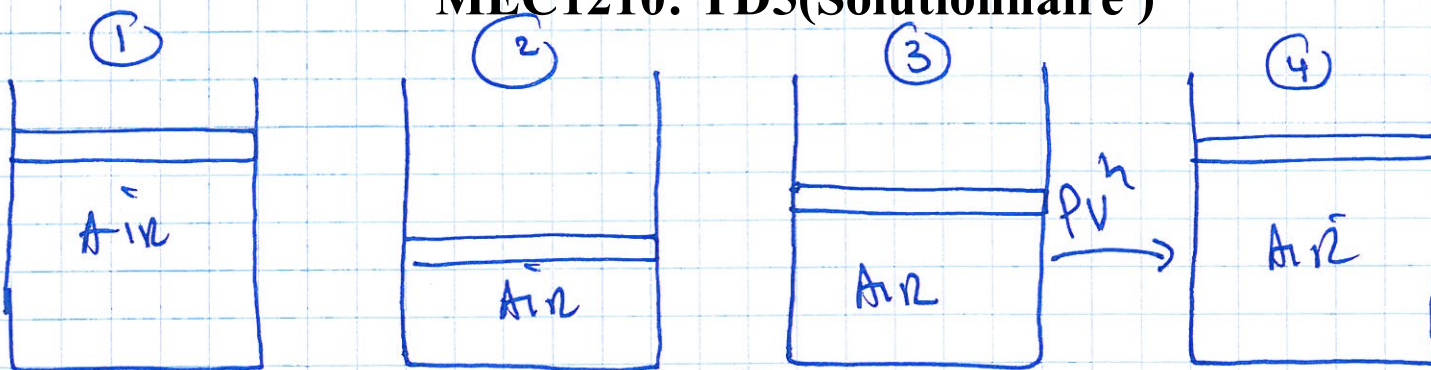


# MEC1210: TD5(Solutionnaire)



$$P_1 = 100 \text{ kPa}$$

$$T_1 = 300 \text{ K}$$

$$S_2 = S_1 \quad q_{in} = 954,7 \text{ kJ/kg}$$

$$P_3 = P_2$$

$$V_2 = \frac{V_1}{11,63377}$$

$$V_4 = V_1$$

$$h = 1,334877$$

Hypothèse:

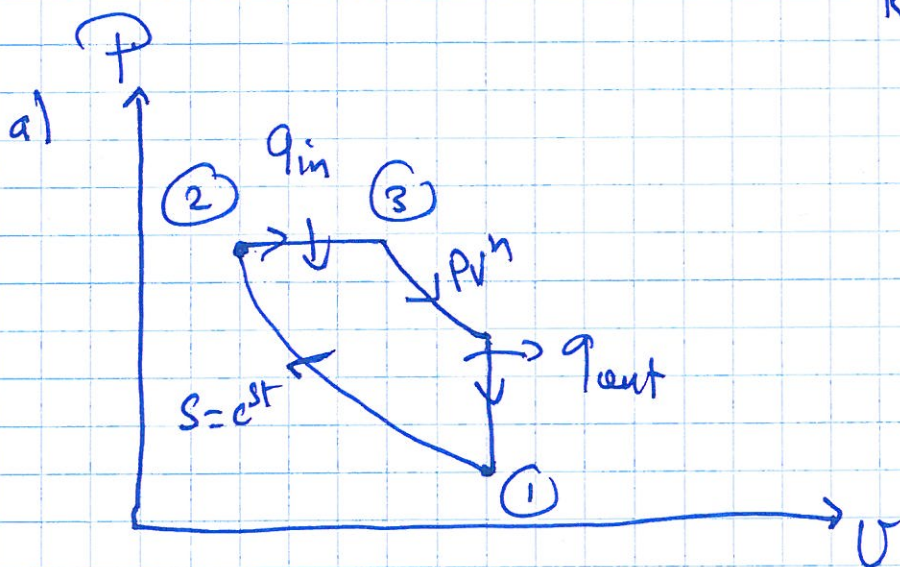
Air gaz parfait à  $c_p$  et  $c_v = c^{stes}$

$$c_v = 0,718 \text{ kJ/kg}\cdot\text{K}$$

$$R = 0,287 \text{ kJ/kg}\cdot\text{K}$$

$$c_p = c_v + R = 1,005 \text{ kJ/kg}\cdot\text{K}$$

$$k = c_p / c_v = 1,41397$$



b) ①  $P_1 = 100 \text{ kPa}$ ,  $T_1 = 300 \text{ K}$ ,  $P_1 v_1 = R T_1$

$$\Rightarrow v_1 = \frac{R T_1}{P_1} = \frac{0,287 \cdot 300}{100}$$

$$v_1 = 0,861 \text{ m}^3/\text{kg}$$

①



$$\textcircled{2} \quad V_2 = \frac{V_1}{11,63379} \quad \text{and} \quad v_2 = \frac{v_1}{11,63379} \Rightarrow v_2 = \frac{0,861}{11,63379} = 0,074 \frac{\text{m}^3}{\text{kg}}$$

$$T_1 v_1^{k-1} = T_2 v_2^{k-1} \Rightarrow T_2 = T_1 \left( \frac{v_1}{v_2} \right)^{k-1}$$

$$T_2 = 300 \left( 11,63379 \right)^{(1,3997-1)} = 800 \text{ K} \quad \boxed{T_2 = 800 \text{ K}}$$

$$P_2 v_2 = R T_2 \Rightarrow \boxed{P_2 = \frac{R T_2}{v_2}} \quad P_2 = \frac{0,287 \cdot 800}{0,074} = 3102,7 \text{ kPa}$$

$$\boxed{P_2 = 3102,7 \text{ kPa}}$$

$$\textcircled{3} \quad \boxed{P_3 = P_2 = 3102,7 \text{ kPa}}$$

$$\text{for } \Delta u_{23} = q_{23} - w_{23} \quad (\text{kJ})$$

$$\Delta u_{23} = q_{23} - w_{23} = q_{23} - P(v_3 - v_2) \quad (\text{kJ/kg})$$

$$u_3 - u_2 + P(v_3 - v_2) = q_{23}$$

$$h_3 - h_2 = c_p (T_3 - T_2) = q_{23}$$

$$\boxed{T_3 = \frac{q_{23}}{c_p} + T_2}$$

$$T_3 = \frac{954,75}{1,005} + 800 = 1750 \text{ K}$$

$$\boxed{T_3 = 1750 \text{ K}}$$

$$\textcircled{4} \quad v_4 = v_1 = 0,861 \text{ m}^3/\text{kg}$$

$$\text{Polytropic } 3-4 \Rightarrow P_4 v_4^\gamma = P_3 v_3^\gamma \Rightarrow$$

$$P_4 = P_3 \cdot \left( \frac{v_3}{v_4} \right)^\gamma$$



$$P_3 V_3 = RT_3 \Rightarrow V_3 = \frac{RT_3}{P_3} \quad V_3 = \frac{0,287 \cdot 1710}{3102,7} = 0,16188 \frac{\text{m}^3}{\text{kg}}$$

$$V_3 = 0,16188 \text{ m}^3/\text{kg}$$

D'ici  $P_4 = 3102,7 \left( \frac{0,16188}{0,861} \right)^{1,334833} = 333,33 \text{ kPa}$

$$P_4 = 333,33 \text{ kPa}$$

$$P_4 V_4 = RT_4 \Rightarrow T_4 = \frac{P_4 V_4}{R}$$

$$T_4 = \frac{333,33 \cdot 0,861}{0,287} = 1000 \text{ K} \quad T_4 = 1000 \text{ K}$$

c)  $W_{12}, W_{23}, W_{34}, W_{41}$   
 1-2 isotherme, 1<sup>er</sup> loi  $\Delta u_{12} = q_{12} - W_{12}$  (Adiabatique)

$$W_{12} = u_1 - u_2 = C_v (T_1 - T_2)$$

$$W_{12} = 0,718 (300 - 600) = -359 \text{ kJ/kg} \quad \text{Sur l'air}$$

Alternative:  $W_{12} = \frac{P_2 V_2 - P_1 V_1}{1-\kappa} = \frac{3102,7 \cdot 0,074 - 100 \cdot 0,861}{1-1,3997} = -359 \text{ kJ/kg}$

2-3  $P = C^{st_2}$

$$W_{23} = \frac{P}{3} (V_3 - V_2)$$

$$W_{23} = 3102,7 (0,16188 - 0,074) = 272,66 \text{ kJ/kg} \quad \text{Par l'air}$$

3-4 Polytropique

$$W_{34} = \frac{P_4 V_4 - P_3 V_3}{1-n} = \frac{333,33 \cdot 0,861 - 3102,7 \cdot 0,16188}{1-1,334833} = 642,83 \text{ kJ/kg}$$

$$W_{34} = +642,83 \text{ kJ/kg} \quad \text{Par l'air}$$

(3)



4-1) volume constant  $W_{41} = 0$

d)  $q_{12}, q_{23}, q_{34}, q_{41}$ .

$q_{12} = 0$  (adiabatique)

$q_{23} = 954,75 \text{ kJ/kg}$  (donnée) à l'air

3-4) à l'air  $\Delta U_{34} = q_{34} - W_{34}$

$$q_{34} = \Delta U_{34} + W_{34} = C_v (T_4 - T_3) + W_{34}$$

$$q_{34} = 0,718 (1000 - 1750) + 642,83 = 104,33 \frac{\text{kJ}}{\text{kg}} \text{ à l'air}$$

4-1) à l'air  $\Delta U_{41} = q_{41} - W_{41} = 0$  ( $V = \text{const}$ )

$$\Rightarrow q_{41} = \Delta U_{41} + W_{41} = C_v (T_1 - T_4) + W_{41}$$

$$q_{41} = 0,718 (300 - 1000) = -502,6 \frac{\text{kJ}}{\text{kg}} \text{ par l'air}$$

e)  $\eta_{th} = ?$

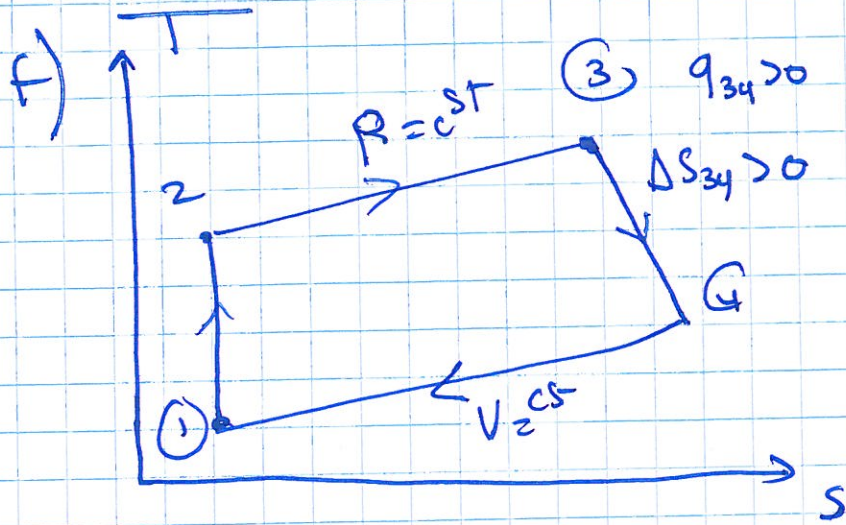
$$\eta_{th} = \frac{W_{\text{net}}}{q_{\text{in}}} = \frac{W_{12} + W_{23} + W_{34} + W_{41}}{q_{23} + q_{34}}$$

$$\eta_{th} = \frac{-359 + 272,66 + 642,83 + 0}{954,75 + 104,33} = 52,54\% \quad \boxed{\eta_{th} = 52,54\%}$$

Alternatif:  $\eta_{th} = 1 - \frac{q_{\text{out}}}{q_{\text{in}}} = 1 - \frac{q_{41}}{q_{23} + q_{34}} = 1 - \frac{502,6}{954,75 + 104,33}$

$$\boxed{\eta_{th} = 52,54\%}$$





$$\Delta S_{34} = C_p \ln \frac{T_4}{T_3} - R \ln \frac{P_4}{P_3}$$

$$= 1005 \cdot \ln \frac{1000}{1780} - 0,287 \ln \frac{333,33}{300,7}$$

$$= 0,078 \frac{kJ}{kg \cdot K} > 0$$

$$S_4 > S_3$$

Refaire b) avec  $C_{p,2}$  et  $C_v$  variables

①  $P_1 = 100 \text{ kPa}$   
 $T_1 = 300 \text{ K} \rightarrow \text{A17 } U_{r,1} = 621,2, \quad v_1 = \frac{R T_1}{P_1} = 0,861$

$$\frac{v_2}{v_1} = \frac{U_{r,2}}{U_{r,1}} \Rightarrow U_{r,2} = \frac{v_2}{v_1} U_{r,1} = \frac{621,2}{11,63335} = 53,38$$

T	$U_r$	h
760	55,54	778,18
X	53,38	h
780	51,64	800,03

$$\Rightarrow X = T_2 = 770,99 \text{ K}$$

$$h_2 = 790,19 \text{ kJ/kg}$$

interpolation

$$\frac{P_1 v_1}{R T_1} = \frac{P_2 v_2}{R T_2} \Rightarrow P_2 = P_1 \frac{v_1 T_2}{v_2 T_1}$$

$$P_2 = 100 \cdot 11,63335 \cdot \frac{770,99}{300}$$

$$P_2 = 2989,85 \text{ kPa}$$



③ 1<sup>er</sup> loi 2-3

$$\Delta u_{23} = q_{23} - w_{23} \Rightarrow \Delta u_{23} = q_{23} - P(u_3 - u_2)$$

$$q_{23} = u_3 - u_2 + P(u_3 - u_2) = h_3 - h_2 \Rightarrow h_3 = q_{23} + h_2$$

$$h_3 = 954,75 + 790,19 = 1744,94 \text{ kJ/kg}$$

$$h_3 = 1744,94 \text{ kJ/kg} \xrightarrow[\text{interpolation.}]{\text{A7}}$$

$$T_3 = 1589,81 \text{ K}$$

$$P_3 = P_2 = 2989,85 \text{ kPa}$$

$$\textcircled{4} P_3 V_3^\gamma = P_4 V_4^\gamma \Rightarrow P_4 = P_3 \left( \frac{V_3}{V_4} \right)^\gamma \quad \gamma = 1,334877$$

$$\text{OR } V_4 = V_1 = 0,861 \text{ m}^3/\text{kg}$$

$$P_3 V_3 = RT_3 \Rightarrow V_3 = \frac{RT_3}{P_3} = \frac{0,287 \cdot 1589,81}{2989,85} = 0,15261 \frac{\text{m}^3}{\text{kg}}$$

$$P_4 = 2989,85 \cdot \left( \frac{0,15261}{0,861} \right)^{1,334877} = 296,89 \text{ kPa}$$

$$P_4 = 296,89 \text{ kPa}$$

$$P_4 V_4 = RT_4 \Rightarrow T_4 = \frac{P_4 V_4}{R}$$

$$T_4 = \frac{296,89 \cdot 0,861}{0,287} = 890,67 \text{ K}$$

$$T_4 = 890,67 \text{ K}$$

⑥